

PII: S0040-4039(97)00657-6

Unprecedented Grob-type Fragmentation of 5-dioxolan-bicyclo[4.2.0]octan-2-ones into 3-(methoxycarbonylmethyl)cyclohexanones

Marzia De Giacomo, Rinaldo Marini Bettolo* and Rita Scarpelli

Dipartimento di Chimica and Centro di Studio del CNR per la Chimica delle Sostanze Organiche Naturali Università degli Studi "La Sapienza", p.le A. Moro 5, I-00185 Roma, Italy

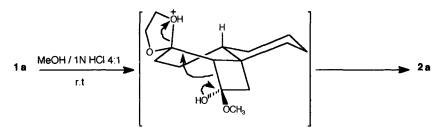
Abstract: The unprecedented Grob-type fragmentations of 5-dioxolan-bicyclo[4,2,0]octan-2-ones I into 3-(methoxycarbonylmethyl)cyclohexanones 2 are described. This fragmentation, in combination with the allene photoannelation used to produce compounds 1, might constitute an indirect way for achieving conjugate addition of methyl acetate; new syntheses of kaurane, stemarane and thirsiflorane diterpenes might be also envisaged from 2c and 2d. © 1997 Elsevier Science Ltd.

In the course of our continuing effort towards the synthesis of tetracyclic diterpenes we found that 5-dioxolan-bicyclo[4.2.0]octan-2-one 1a¹⁻³ is quantitatively converted into 3-(methoxycarbonylmethyl)cyclohexanone 2a⁴ by the action of a 4:1 mixture of MeOH/1N HCl at r.t. for 2h. The reaction proceeds cleanly and the corresponding bicyclo[4.2.0]octan-2,8-dione 3 and 9-(carboxymethyl)-2-decalone 4 were not detected in the course of the reaction. The same unprecedented outcome was recorded by us when this reaction was applied to 1b⁵, 1c⁶ and 1d⁷ from which 2b⁸, 2c⁹ and 2d¹⁰ were obtained respectively.

The conversion of compounds 1 into 2, which gives rise to two C-atoms at a different level of oxidation, is unique in the sense that the cyclobutane and dioxolane rings are opened "simultaneously". These Grob-type 11b-c fragmentations appear to occur because of the peculiar *antiperiplanar* arrangement, in the emiacetalic intermediate, of the CH-C(OH)OCH3 cyclobutane bond and one of the dioxolane C-O bond, as described for the conversion of 1a into 2a in the scheme.

This fragmentation is, in our opinion, quite interesting, not only because is novel and constitutes therefore a further variant of these processes, but also because compounds of type 1 are efficiently available by stereoselective allene photoaddition to the proper α,β -unsaturated ketone and simple functionality

manipulations; the whole sequence might be therefore very useful, in alternative to the various conjugate addition methodologies, for the introduction of an angular methyl acetate side chain. Finally 2c and 2d are precursors of 3-(formylmethyl)cyclohexan-1-one intermediates from which new syntheses of kaurane, stemarane and thirsiflorane diterpenes, by intramolecular reductive coupling 12 and rearrangement 13 of the cyclization products, might be envisaged.



Scheme. A plausible reaction pathway for the conversion of 1a into 2a

Acknowledgments

This work was supported by the Ministero dell'Università e Ricerca Scientifica e Tecnologica (MURST).

References and Notes

- The elemental composition of all compounds, described in this communication, has been determined by elemental analysis or high resolution mass spectrometry; IR in cm⁻¹, NMR in δ.
- IR (CCl₄): 1776; 1 H-NMR (CDCl₃): 2.35 (dd, J = 5.5, 18.5 Hz, 1H), 2.60-2.70 (m, 1H), 2.97 (dd, 2. $J = 3.0, 18.5 \text{ Hz}, 1H), 3.70-3.90 \text{ (m, 4H); } ^{13}\text{C-NMR} \text{ (CDCl}_3): } 22.8, 24.3, 25.9, 29.5, 32.0, 35.4,$ 37.3, 38.5, 49.3, 64.0, 64.4, 70.0, 108.0, 206.1; colourless oil.
- Obtained from the corresponding olefin as described for the preparation of 1d in ref.7. 3.
- TLC (SiO₂): (petroleum ether (40-70°)/Et₂O) 6:4. $R_{/2a} < R_{/1a}$; IR (CCl₄): 1734, 1717; ¹H-NMR (CDCl₃): 3.61 (s, 3H); ¹³C-NMR (CDCl₃): 20.5, 25.8, 27.3, 28.3, 32.7, 37.0, 39.6, 41.0, 44.5, 51.1, 52.7, 172.1, 210.8; low melting oil (38-39°C).

 A. Bartoletti, M. Berettoni, F. Catteruccia, G. De Chiara, R. Marini Bettolo, C. Mastrangeli, R.
- 5. Scarpelli, C. Tozzi, D. Lamba, Gazz. Chim. Ital., 1996, 126, 223.
- 6. Derived from (+)-podocarpic acid; IR (CCl₄): 1774; ¹H-NMR (CDCl₃): 0.80 (s, 3H), 0.82 (s, 3H), 1.01 (s, 3H), 3.80-4.00 (m, 4H); ¹³C-NMR (CDCl₃): 15.5, 18.0, 21.0, 21.7, 24.5, 30.0, 30.4, 30.5, 31.3, 32.9, 33.0, 39.0, 42.0, 45.6, 46.1, 47.5, 63.2, 64.2, 64.4, 108.8, 208.4; m.p. (petroleum ether (40-70°)/Et₂O): 99.5-100.6°C.
- 7. Derived from (-)-abietic acid; M. Berettoni, G. De Chiara, T. Iacoangeli, P. Lo Surdo, R. Marini Bettolo, L. Montagnini di Mirabello, L. Nicolini, R. Scarpelli, Helv. Chim. Acta, 1996, 79, 2035.
- 8. TLC (SiO_2): (petroleum ether (40-70°)/Et₂O) 6:4; $R_{f2b} < R_{f1b}$; IR (CCl₄): 1734, 1711, as epimeric mixture at C(9).
- TLC (SiO₂): (petroleum ether (40-70°)/Et₂O) 1:1, $R_{f2c} > R_{f1c}$; IR (CCl₄): 1711, 1733; ¹H-NMR (CDCl₃): 0.85 (s, 6H) 1.05 (s, 3H), 3.55 (s, 3H); ¹³C-NMR (CDCl₃): 15.8, 18.3, 21.4, 21.7, 28.3, 9. 28.6, 29.5, 30.1, 32.9, 33.0, 34.0, 34.3, 37.9, 40.3, 41.5, 41.9, 44.2, 45.3, 47.3, 173.7, 211.8; colourless oil.
- 10. TLC (\$iO₂): (petroleum ether (40-70°)/Et₂O) 4:6, R_{f2d} > R_{f1d}; IR (CCl₄): 1735; ¹H-NMR (CDCl₃): 0.82 (s, 3H), 1.08 (s, 3H), 3.54 (s, 3H), 3.57 (s, 3H); ¹³C-NMR (CDCl₃): 16.0, 16.2, 17.3, 20.4, 20.9, 35.6, 36.5, 36.9, 38.5, 38.9, 40.7, 40.8, 47.4, 51.1, 51.7, 54.4, 57.5, 172.0, 179.1, 210.1; m.p. (petroleum ether (40-70°)/Et₂O): 136.8-137.1°C.
- 11. a) P. Deslongchamps, Stereoelectronic Effects in Organic Chemistry, Pergamon Press, Oxford, 1983, p. 273; b) C.A. Grob, Angew. Chem. Int. Ed. Engl. 1969, 8, 535; c) Fragmentation Reactions, P. Weyerstahl and H. Marschall in Comprehensive Organic Synthesis, Ed.-in-Chief B.M. Trost, Vol. Ed. E. Winterfeldt, vol. 6, Pergamon Press, Oxford, 1991, p. 1041.
- a) E.J. Corey, R.L. Danheiser, S. Chandrasekaran, J. Org. Chem., 1976, 41, 260; b) J.L. Namy, J. Souppe, H.B. Kagan, Tetrahedron Lett., 1983, 24, 765; c) G.A. Molander, C.J. Kenny, J. Org. Chem., 1988, 53, 2132.
- Natural Products Chemistry, eds. K. Nakanishi, T. Goto, S. Itô, S. Natori, S. Nozoe, vol. 1, Academic Press, Inc., New York, 1974, p. 255.